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ABSTRACT BOOK

Illuminating a Solvent-Dependent Hierarchy for Aromatic CH/ π Complexes with Dynamic Covalent Glyco-Balances

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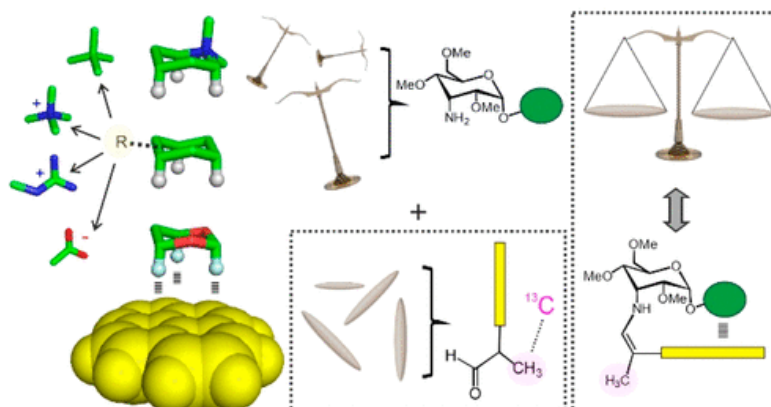
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CH/ π interactions are prevalent among aromatic complexes and represent invaluable tools for stabilizing well-defined molecular architectures. Typically, their energy contributions are exceptionally sensitive to structural and environmental factors, resulting in a context-dependent nature that has led to conflicting findings in the scientific literature. Consequently, a universally accepted hierarchy for aromatic CH/ π interactions has remained elusive. Herein, we present a comprehensive experimental investigation of aromatic CH/ π complexes, employing a novel approach that involves isotopically labeled glyco-balances generated *in situ*. This innovative strategy allows us to uncover both thermodynamic and kinetic information. Our analyses yielded more than 180 new free energy values while considering key factors such as solvent properties, the interaction geometry, and the presence and nature of accompanying counterions. Remarkably, our results challenge conventional wisdom regarding the stability order of common aromatic complexes and provide valuable guidelines for the selection of optimal interacting partners in every chemical environment, allowing the design of tailored aromatic complexes with applications in supramolecular chemistry, organocatalysis, and/or material sciences.¹



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References:

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